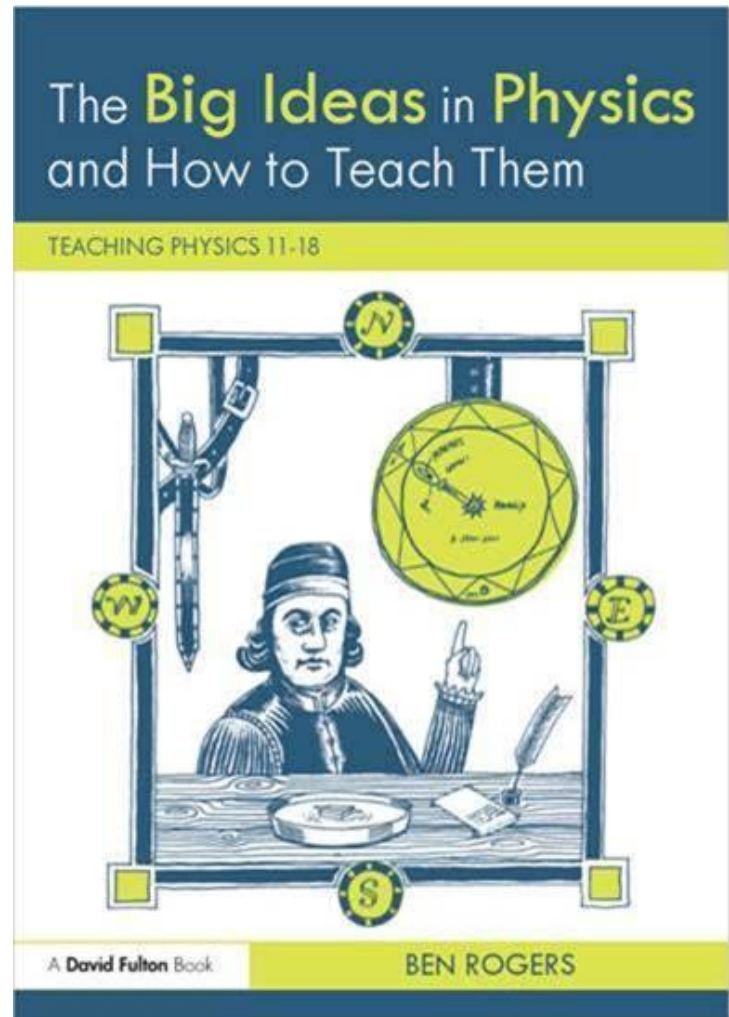


A physicist is the sum of  
the problems she can  
solve.

Ben Rogers



## Part #1 Problem solving is knowledge

Great physicists are not born, they are made,  
problem by problem.

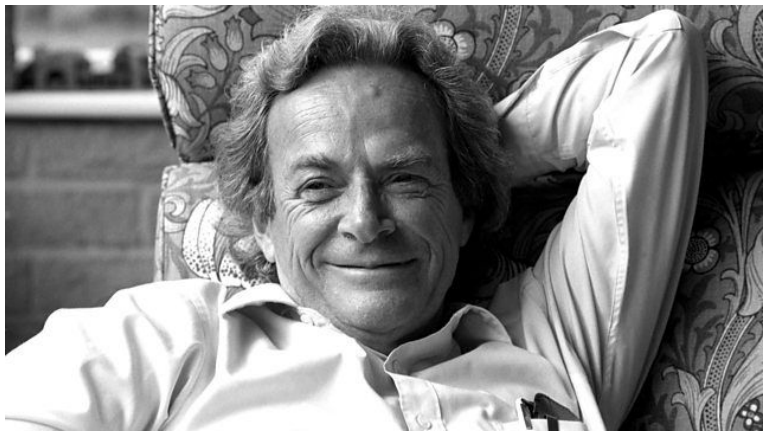
..the science student, confronted with a problem, seeks to see it as like one or more of the exemplary problems he has encountered before.

Thomas Kuhn 1977: 297

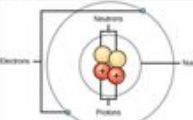
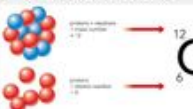

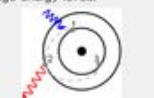
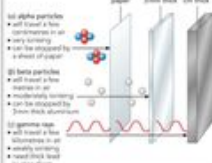


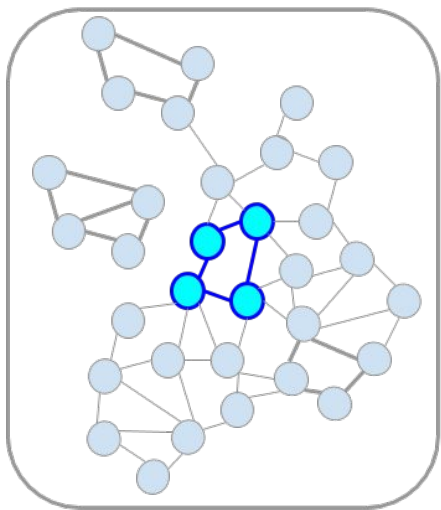
*Know how to solve every problem that has been solved.*

Richard Feynman

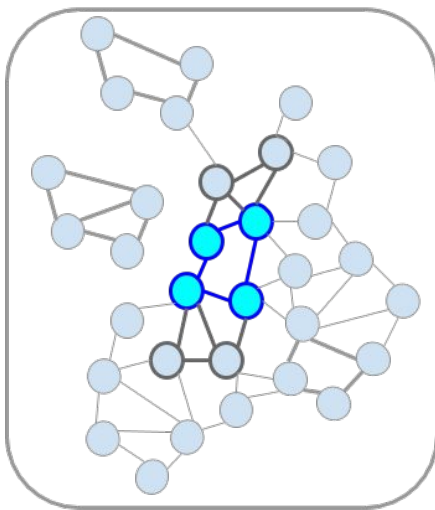


# From Knowledge Organisers to Problem Solving

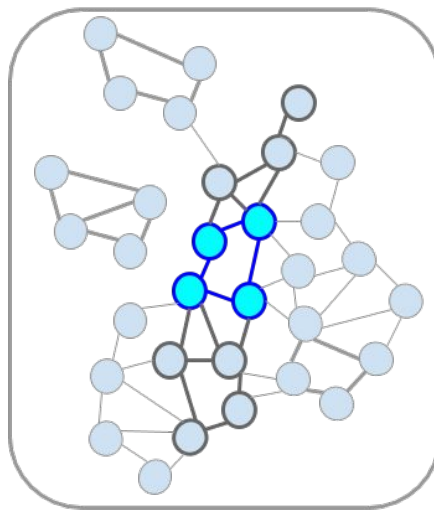
<p>Atoms are made up of smaller particles called <b>protons</b>, <b>neutrons</b> and <b>electrons</b>.</p>  <ul style="list-style-type: none"> <li>The <b>nucleus</b> is the centre of the atom made from neutrons and protons bonded together</li> <li><b>Protons</b> have a mass of 1 and a charge of +1 (positive)</li> <li><b>Neutrons</b> have a mass of 1 and a charge of 0 (no charge)</li> <li><b>Electrons</b> have a mass of 0 (no mass) and a charge of -1 (negative)</li> </ul> <p>An atom always has an equal number of protons and electrons so it has no overall charge (0). Atoms of the same element have the same number of protons - for example all carbon atoms have 6 protons. Some atoms have different numbers of neutrons - these are called <b>isotopes</b>.</p> <p>Isotopes of an element have the same number of protons but different numbers of neutrons.</p> 	<p>When using chemical symbols the larger number is the <b>mass number</b>. This tells us the number of protons and neutrons. The smaller number tells us the number of protons. This is the <b>atomic number</b>. This also tells us the number of electrons.</p> <p>The number of neutrons can be worked out by doing the mass minus the atomic number (big number minus small number). E.g. Carbon has a mass of 12 and an atomic number of 6. Therefore it has:</p> <ul style="list-style-type: none"> <li>6 protons</li> <li>6 electrons</li> <li>6 neutrons (12-6=6)</li> </ul> <p>The electrons of an atom orbit around the nucleus in energy levels. These energy levels are specific distances from the nucleus.</p>  <p>Electrons can be knocked off by ionising radiation. This causes the atom to become <b>charged</b> and is now called an <b>ion</b>. Ions are atoms with a charge.</p> <p>When electrons absorb or emit electromagnetic radiation they can change energy levels.</p> 	<p>Some unstable nuclei will undergo radioactive decay - these are called <b>radioisotopes</b>. The amount of radioactivity is measured in <b>Becquerel's (Bq)</b>. The decay of radioisotopes is a <b>random</b> process which means it is impossible to predict when a nucleus will decay and emit radiation. There are four types of nuclear radiation:</p> <p><b>Nuclear radiation</b> is where high speed neutrons are released from the nucleus.</p> <p><b>Alpha particles</b> are the largest radiation particles (2 protons, 2 neutrons) so they have a very low penetration, stopped by a few centimetres of air. They are also the most ionising and have +ve charge.</p> <p><b>Beta particles</b> are electrons and are less ionising. They have a -ve charge.</p> <p><b>Gamma particles</b> have no charge and are a form of radiation. They have the most penetration of any radiation (several metres of concrete) and travel at the speed of light. They are the least ionising radiation.</p>  <p>an alpha particle • will travel a few centimetres in air • easily ionising • can be stopped by a sheet of paper</p> <p>a beta particle • will travel a few metres in air • moderately ionising • can be stopped by a thin sheet of aluminium</p> <p>gamma rays • will travel a few kilometres in air • relatively ionising • need thick lead to stop them</p>	<p>Background radiation is the ionising radiation this is around us all the time. This radiation can come from:</p> <ul style="list-style-type: none"> <li><b>Radon gas</b></li> <li>Rocks and soil</li> <li>Cosmic rays</li> <li>Medical uses (X-rays etc)</li> </ul> <p>Some of these sources are natural, others are man-made.</p> <p>The amount of background radiation varies from place to place. It is important to measure the background radiation before you measure the radiation of a radioactive source so you know exactly how much radiation comes from the source and how much is from background radiation. A radiation dose is measured in Sieverts (Sv).</p> <p>We can represent nuclear decay using nuclear equations.</p> <p>Alpha decay is the same as having a helium nucleus emitted:</p> ${}^{235}_{92}\text{U} \rightarrow {}^{231}_{90}\text{Th} + {}^4_2\text{He}$ <ul style="list-style-type: none"> <li>Check the mass numbers add up correctly (235 = 231 + 4)</li> <li>Check the atomic numbers add up (92 = 90 + 2)</li> </ul> <p>During beta decay an electron is emitted:</p> ${}^{231}_{90}\text{Th} \rightarrow {}^{231}_{91}\text{Pa} + {}^0_{-1}\text{e}$ <p>In this case the proton number increases by one to ensure the atomic numbers balance on either side of the equation. This means the number of protons has changed and therefore the</p>
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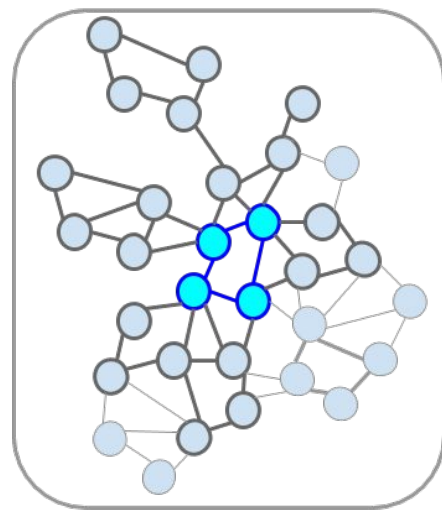
know



understand



use

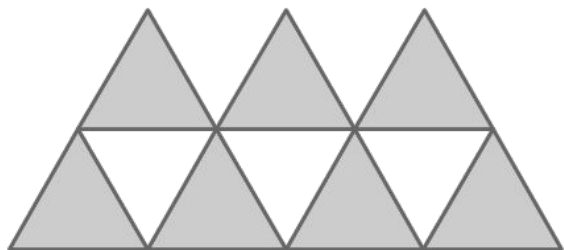
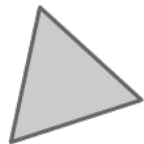


master

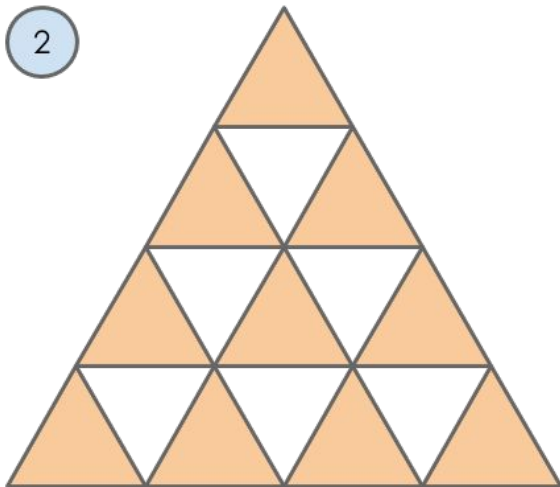


Efrat Furst

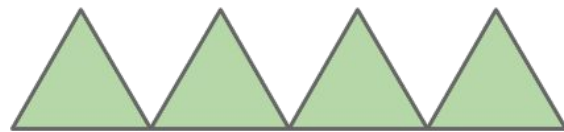
1



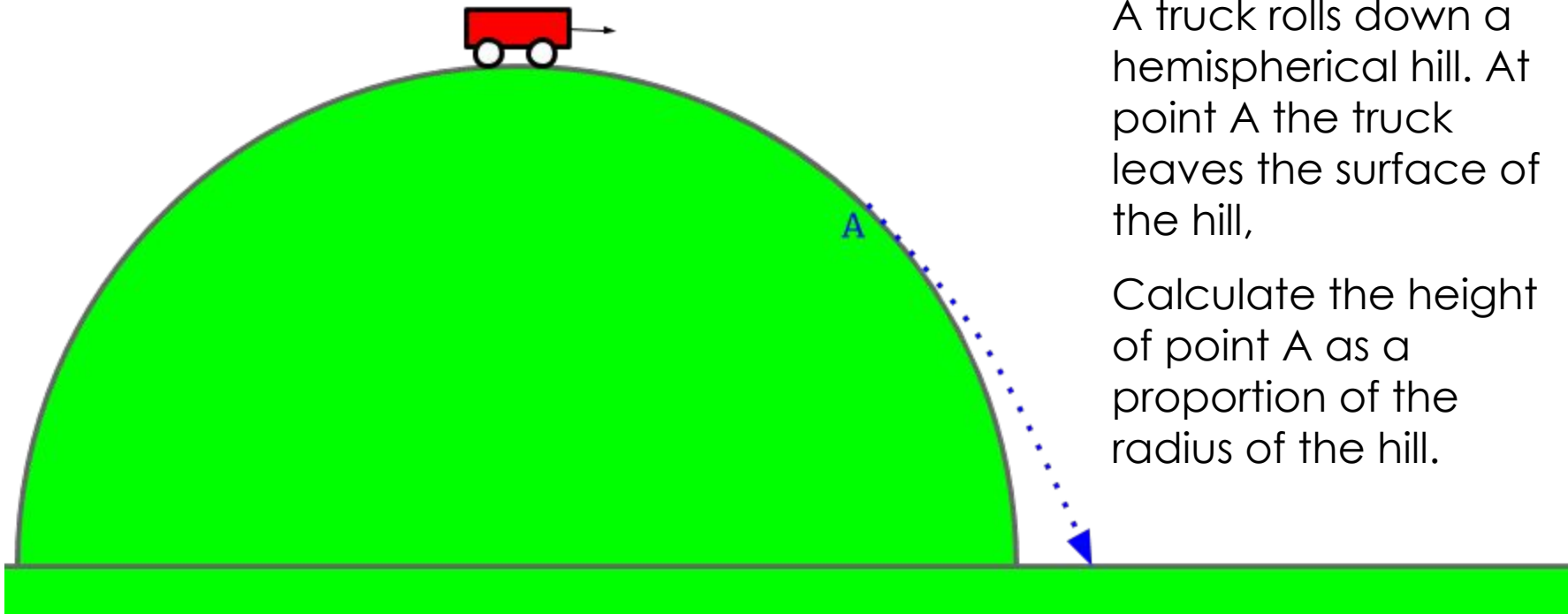
2



3



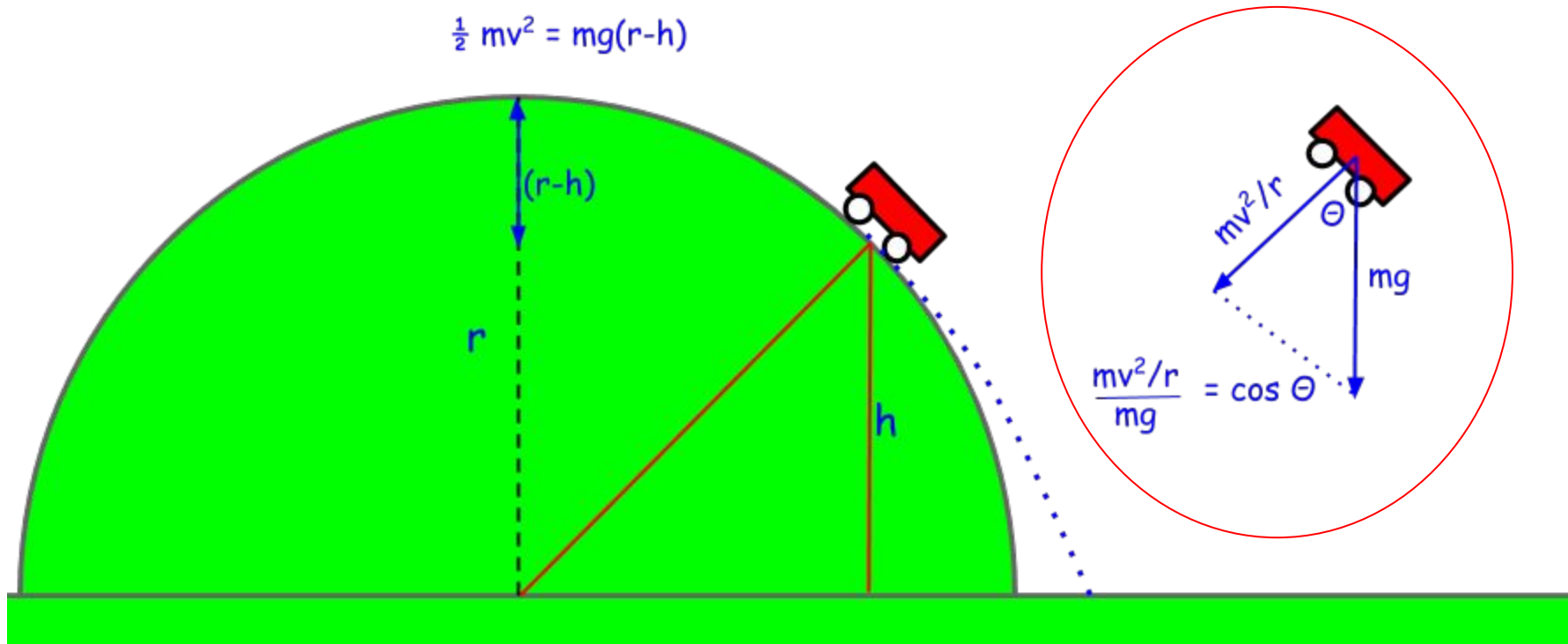
Efrat Furst



A truck rolls down a hemispherical hill. At point A the truck leaves the surface of the hill,

Calculate the height of point A as a proportion of the radius of the hill.





Motion in a circle ( $F=mv^2/r$ )

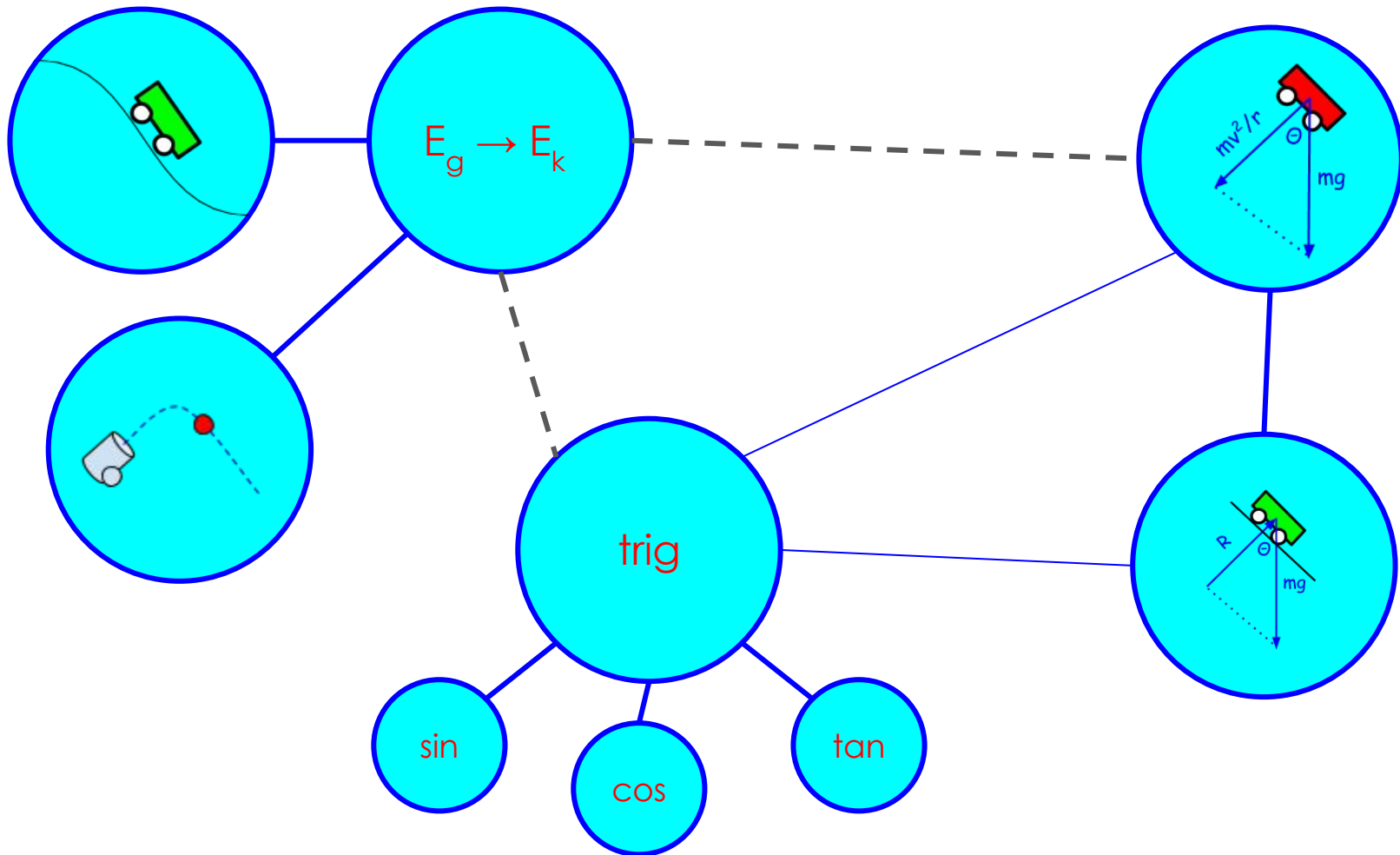
$E_g \rightarrow E_k$

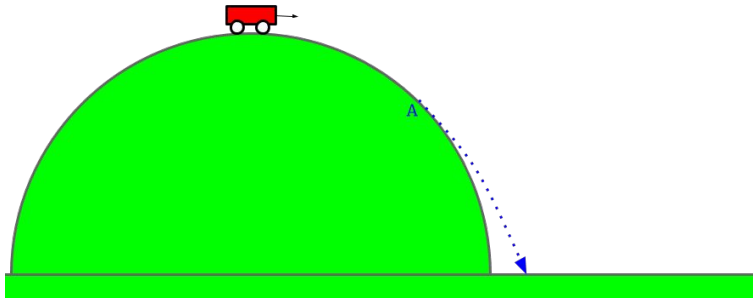
Trigonometry

Ignore friction

Ignore the energy stored in KE of the wheels.

~~Sumat~~





..the science student, confronted with a problem, seeks to see it as like one or more of the exemplary problems he has encountered before.

Always ask extension questions:

- What would happen if the hill was on the Moon?
- What would happen if the hill was straight/parabolic?
- What would happen if you considered friction?
- What would happen if you considered the angular momentum of the wheels?
- What would happen if you used a ball?

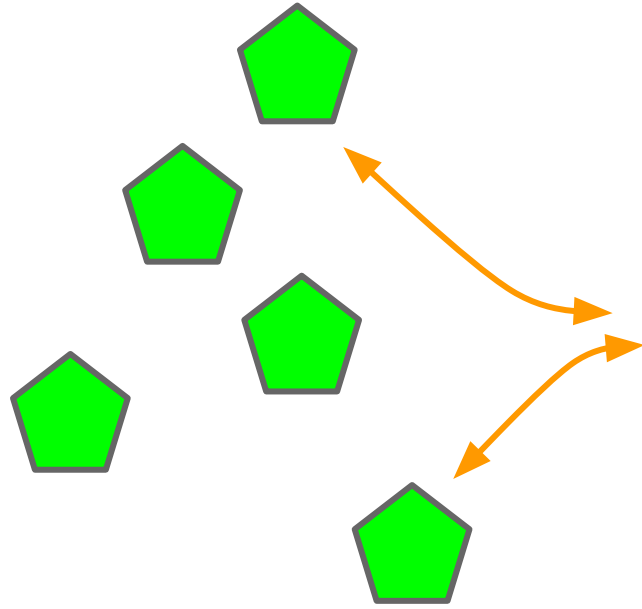
## Part #2 Why knowledge is important for problem solving: Cognitive Load Theory

# Cognitive Load Theory

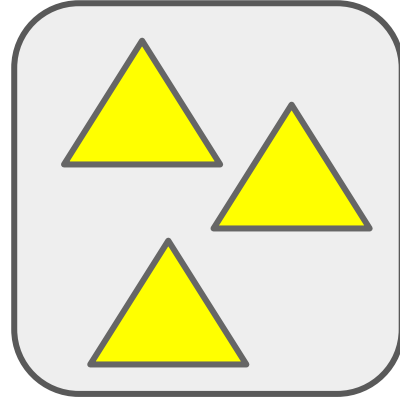
1. We solve problems using working memory.
2. Working memory can hold very little new knowledge.
3. Working memory can draw on long-term memory with very little effort.
4. When working memory is 'full', we cannot make changes to working memory.

John Sweller

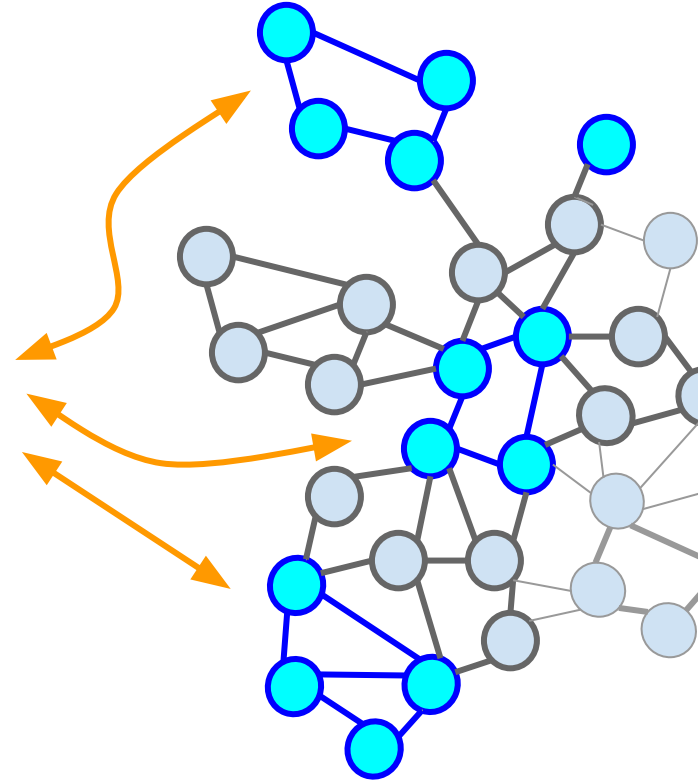




External  
memory



Working  
memory



Schema in  
long-term  
memory

**Takeaway:** reducing Cognitive Load allows more efficient transfer of knowledge to long-term memory.

## Part #3 Reducing the Cognitive Load of a problem using bar-model





... the enormous usefulness of mathematics in the natural sciences is something bordering on the mysterious and that there is no rational explanation for it.

Eugene Wigner



Mathematics is the language in which God has  
written the universe

Galileo Galilei





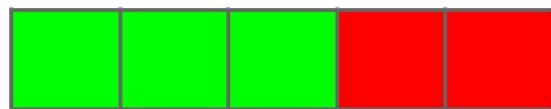
concrete



concrete



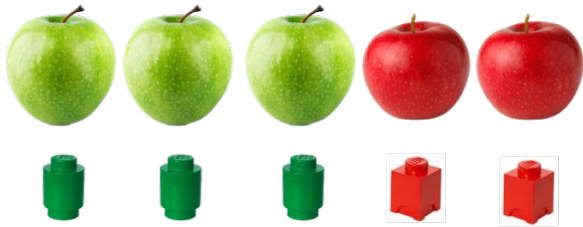
pictorial



pictorial

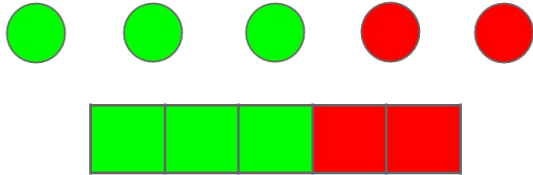
$$3 + 2 = 5$$

abstract



concrete

concrete



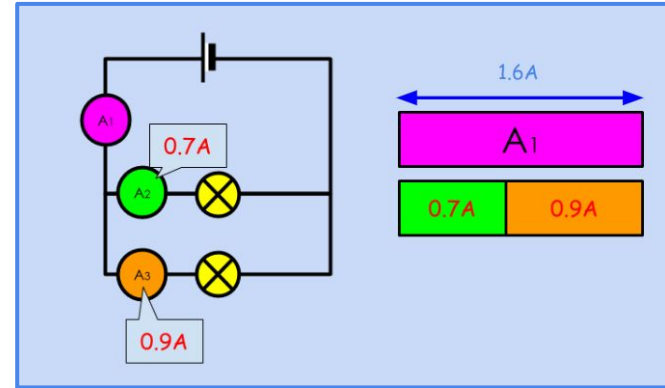
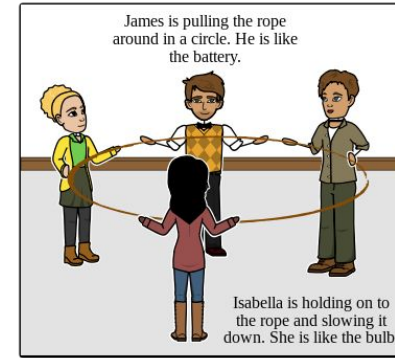
pictorial

pictorial

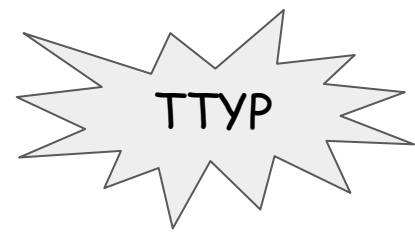
$$3 + 2 = 5$$

abstract

## THE ROPE MODEL



$$A_1 = A_2 + A_3$$



What is the most common **incorrect** answer?

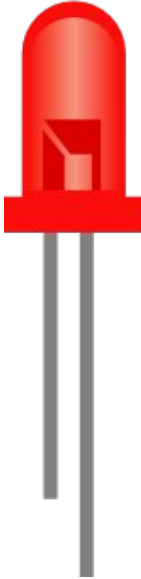
An LED is 80% efficient. If it produces 100J of light, how much electrical energy did it use?

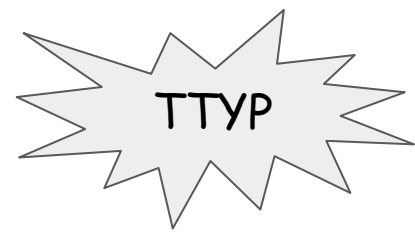
A: 75J

B: 80J

C: 120J

D: 125J





What is the most common **incorrect** answer?

An LED is 80% efficient. If it produces 100J of light, how much electrical energy did it use?

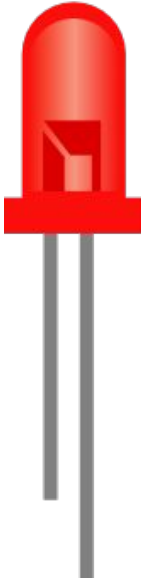
A: 75J

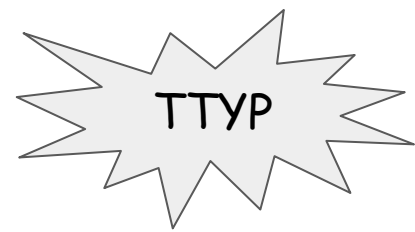
B: 80J

C: 120J



D: 125J





How does this bar-model help your learners?

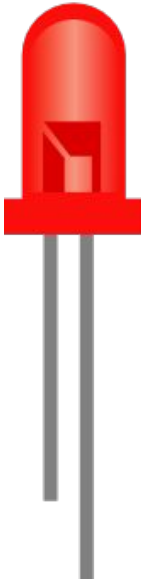
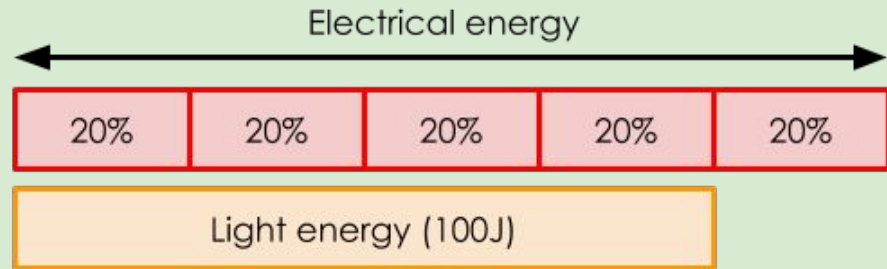
An LED is 80% efficient. If it produces 100J of light, how much electrical energy did it use?

A: 75J

B: 80J

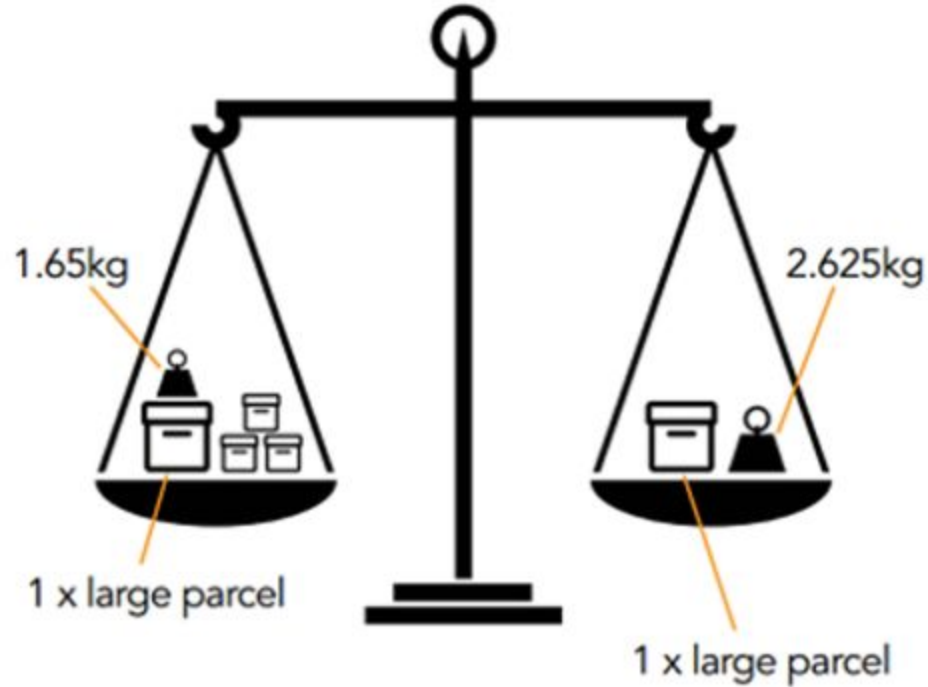
C: 120J

D: 125J



This diagram shows some parcels on a balance scale. Each small parcel is identical.

Calculate the mass of **one** small parcel, in grammes.

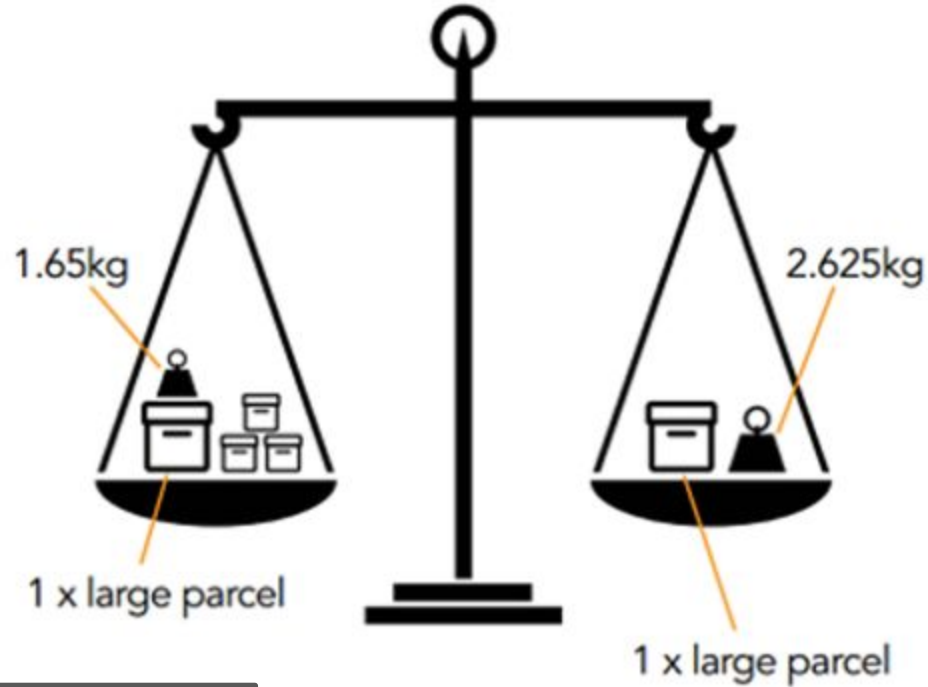


KS2 SATs  
question



This diagram shows some parcels on a balance scale. Each small parcel is identical.

Calculate the mass of **one** small parcel, in grammes.



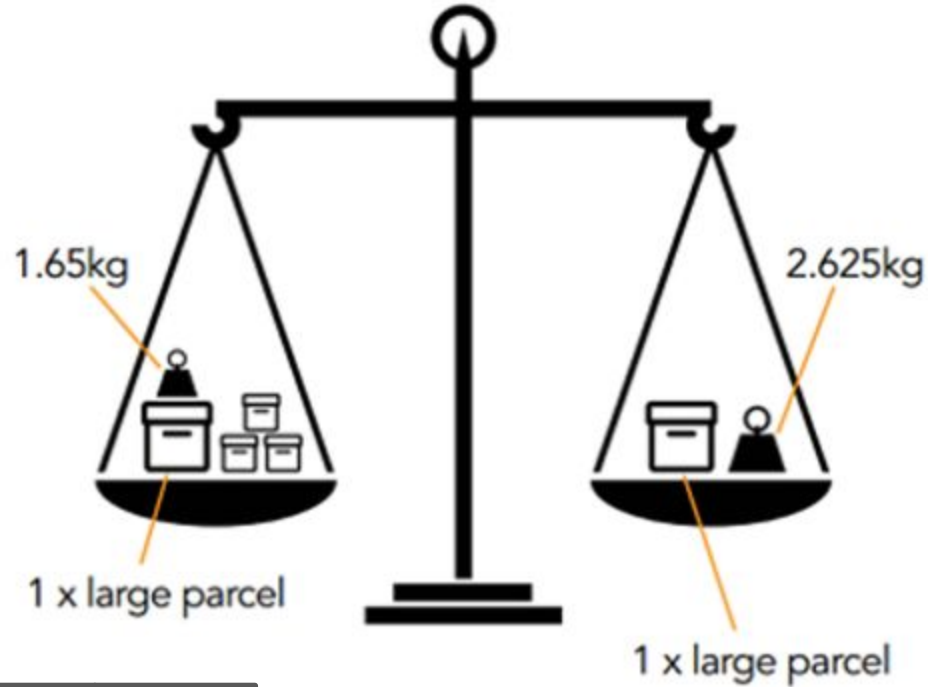
left side

right side

KS2 SATs  
question

This diagram shows some parcels on a balance scale. Each small parcel is identical.

Calculate the mass of **one** small parcel, in grammes.



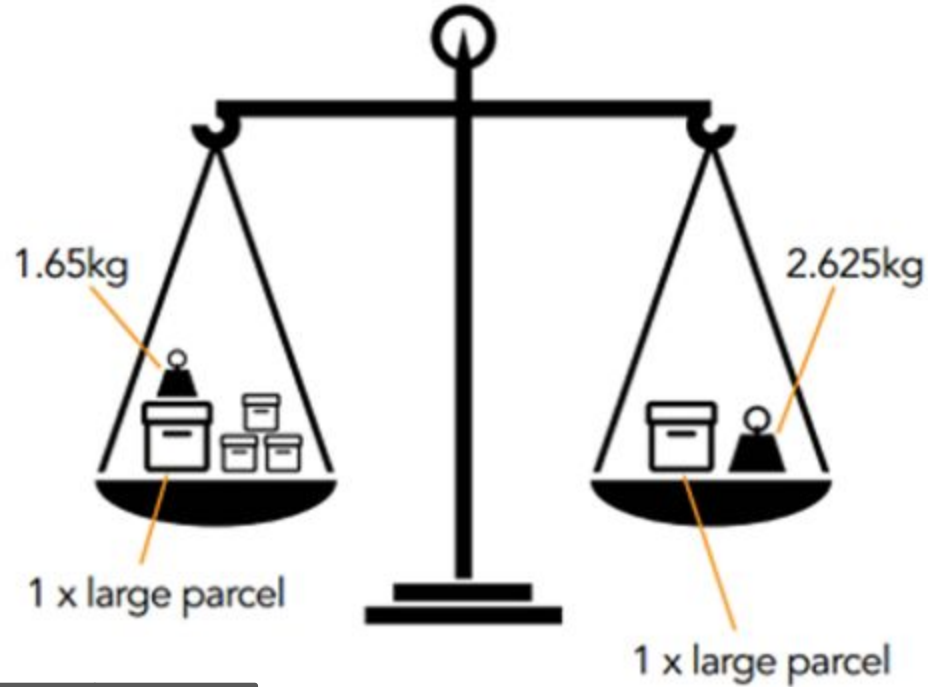
large parcel	1.65kg	small parcel	small parcel	small parcel
--------------	--------	--------------	--------------	--------------

right side



This diagram shows some parcels on a balance scale. Each small parcel is identical.

Calculate the mass of **one** small parcel, in grammes.



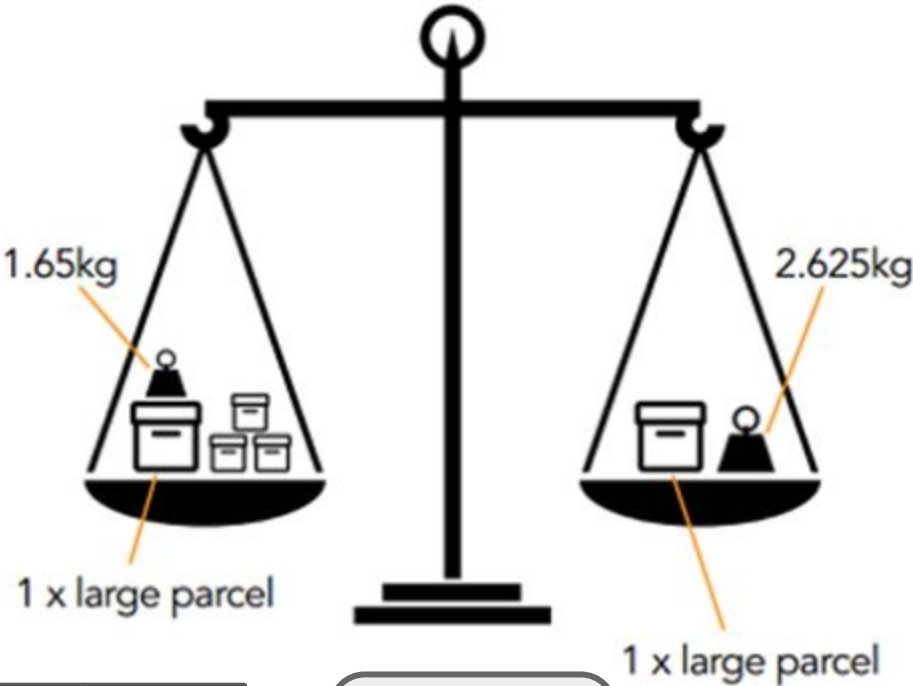
large parcel	1.65kg	small parcel	small parcel	small parcel
--------------	--------	--------------	--------------	--------------

large parcel	2.625kg
--------------	---------

KS2 SATs  
question

This diagram shows some parcels on a balance scale. Each small parcel is identical.

Calculate the mass of **one** small parcel, in grammes.



large parcel	1.65kg
--------------	--------

large parcel	1.65kg
--------------	--------

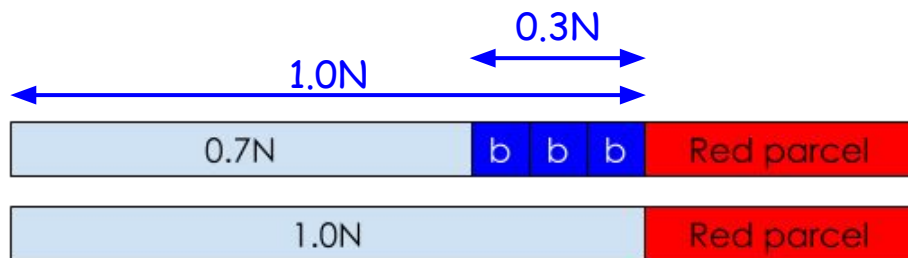
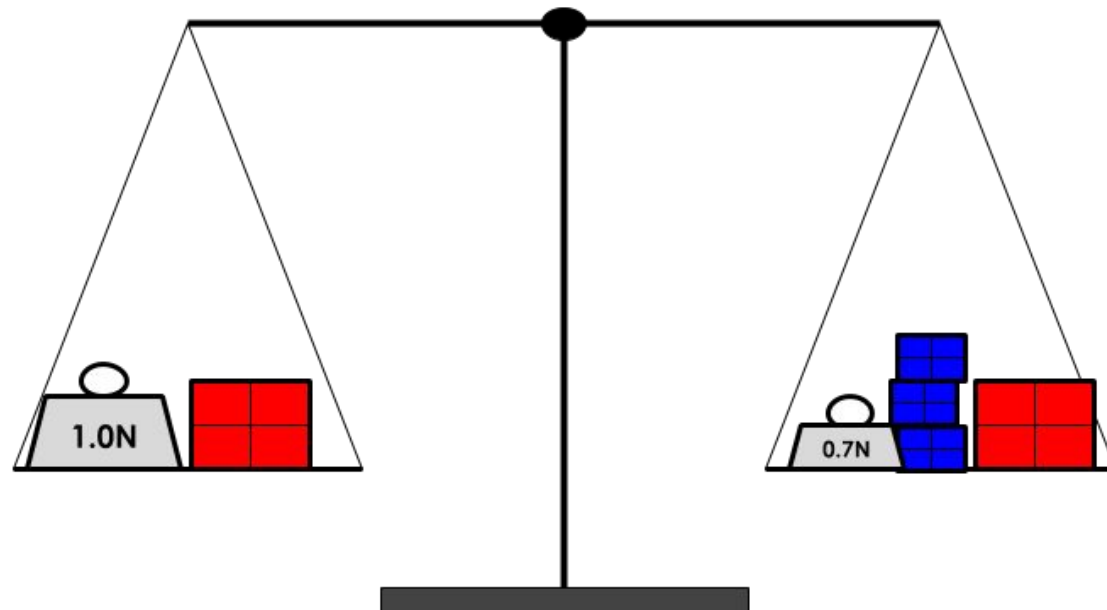
small parcel	small parcel	small parcel
--------------	--------------	--------------

$2.625\text{kg} - 1.65\text{kg} = 0.975\text{kg}$
---

$=0.975/3$   
 $=0.325\text{kg}$

KS2 SATs question

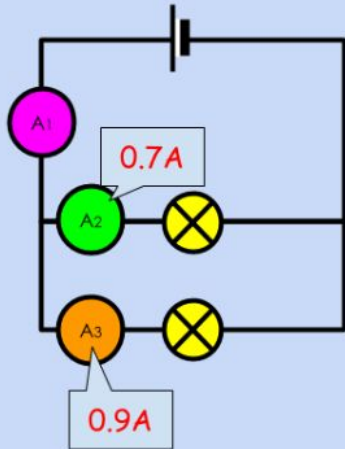
Calculate the weight of each small blue parcel.



1 blue parcel weighs  $0.1\text{N}$

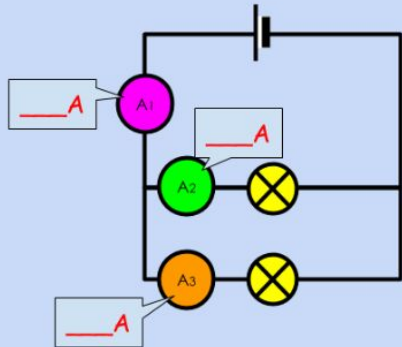
# Electricity

I do



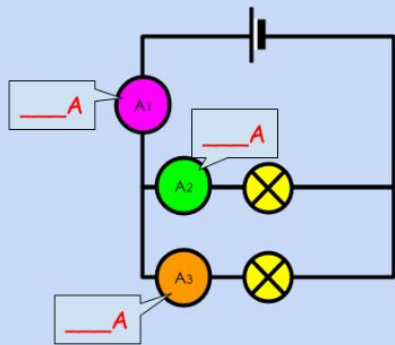
we  
do

If the current through  $A_2$  is 1.2A and the current through  $A_3$  is 0.8A, what is the current through  $A_1$ ?



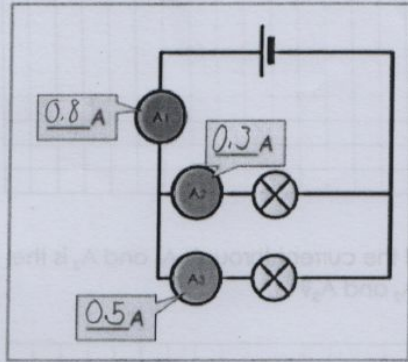
you  
do

If the current through  $A_2$  is 1.0A and the current through  $A_3$  is 0.9A, what is the current through  $A_1$ ?





1. If the current through  $A_2$  is 0.3A and the current through  $A_3$  is 0.5A, what is the current through  $A_1$ ?



$$A_1 = 0.8$$

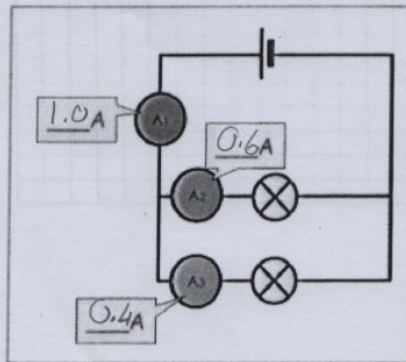
$$A_2 = 0.3$$

$$A_3 = 0.5$$

0.8
0.3 0.5

①

2. If the current through  $A_1$  is 1.0A and the current through  $A_2$  is 0.6A, what is the current through  $A_3$ ?



$$A_1 = 1.0$$

$$A_2 = 0.6$$

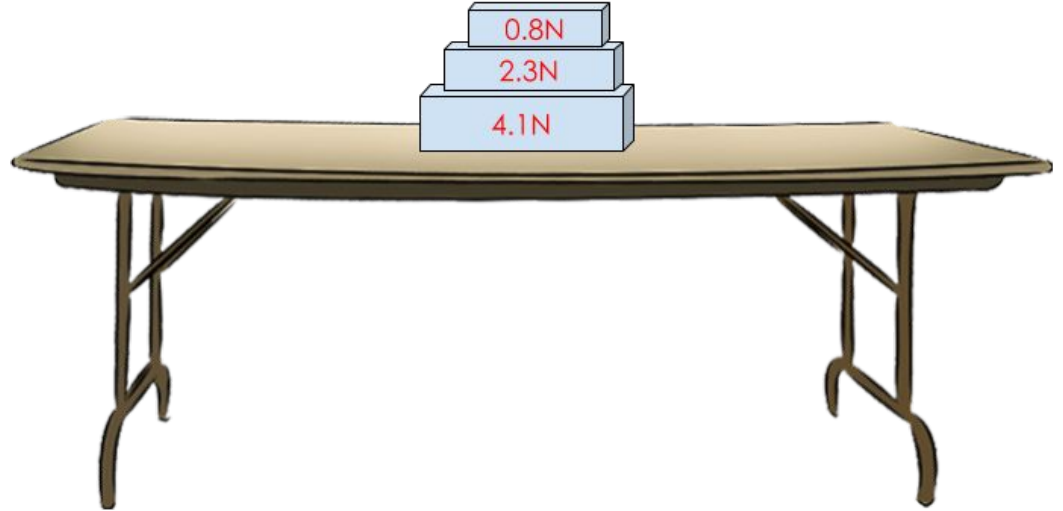
$$A_3 = 0.4$$

1.0
0.6 0.4

①

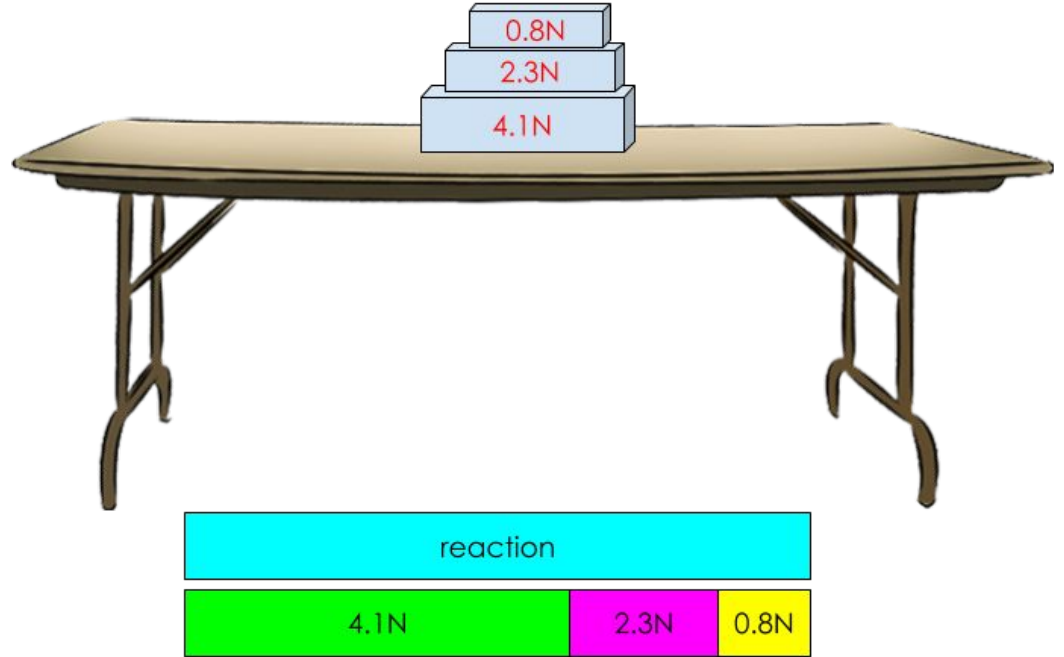
# Applications

Calculate the reaction force from the table.



# Applications

Calculate the reaction force from the table.



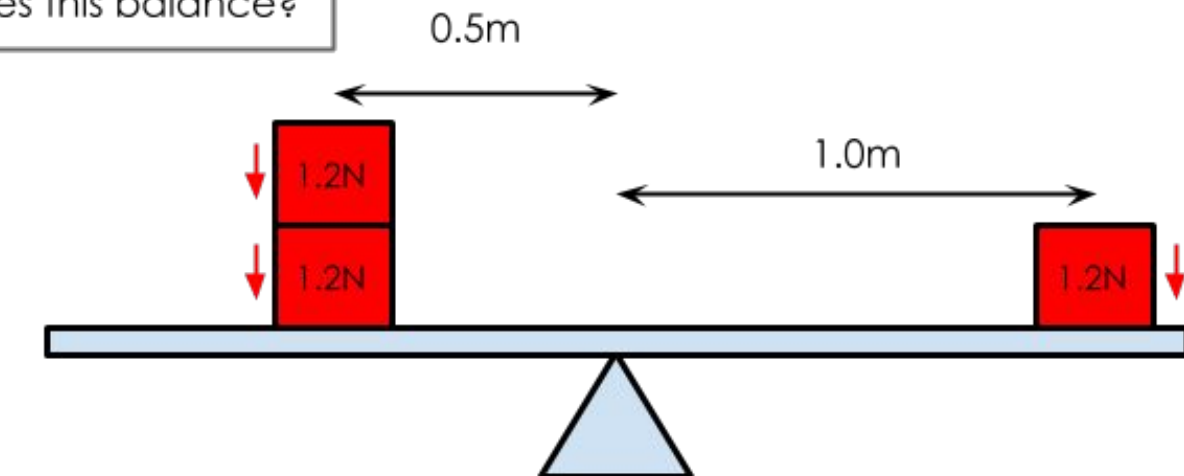
3 friends play tug of war against a teacher. The forces are balanced. The teacher pulls with 600N. One friend pulls with 190N. Another pulls with 290N. How much force does the third friend pull with?



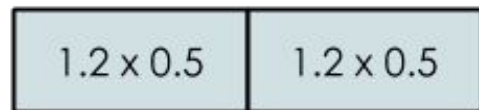
600N
------

190N	290N	?
------	------	---

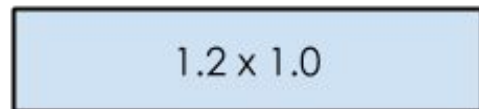
Does this balance?



0.6 Nm    0.6 Nm



2 parcels 0.5m from the pivot:  
Moment =  $2 \times (1.2 \times 0.5) = 2 \times 0.6 = \underline{1.2 \text{ Nm}}$



1 parcel 1.0m from the pivot:  
Moment =  $1.2 \times 1.0 = \underline{1.2 \text{ Nm}}$

1.2 Nm

A teacher uses a Geiger-Müller tube and a counter to measure background radiation. The reading on the counter tube is 34 counts per minute.

The teacher puts a source of beta radiation 15 cm in front of the same Geiger-Müller tube. The reading on the counter tube is now 468 counts per minute.

Calculate how much radiation detected by the Geiger-Müller tube comes from the source of beta radiation.

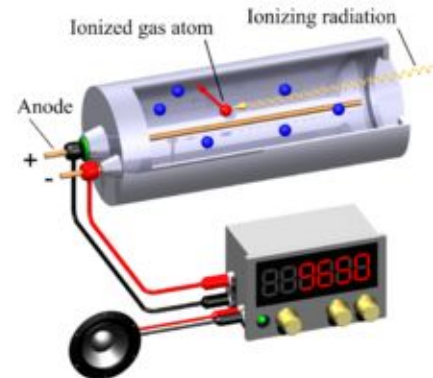
[1 mark]

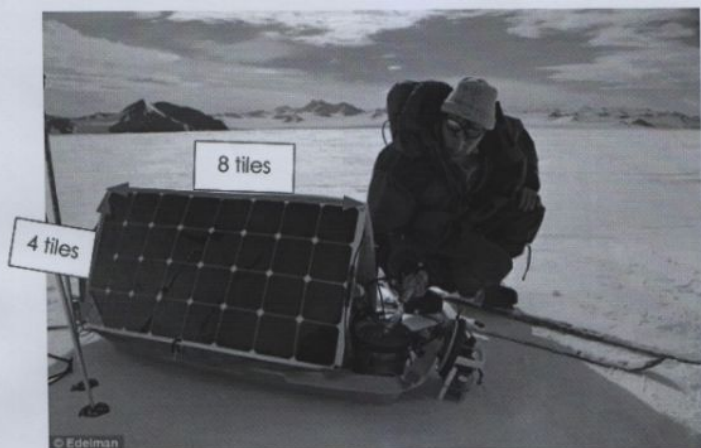
[<https://www.bbc.com/education/guides/z2n8h39/revision/2>]

measurement from the counter

source

background



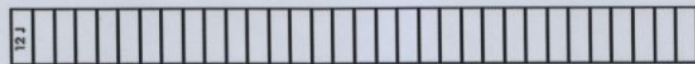


Each second 12J of energy from the Sun reaches each tile.

Calculate the total amount of energy reaching the solar panel in 1 second.

[3]

Number of tiles =  $4 \times 8 = 32$  .....

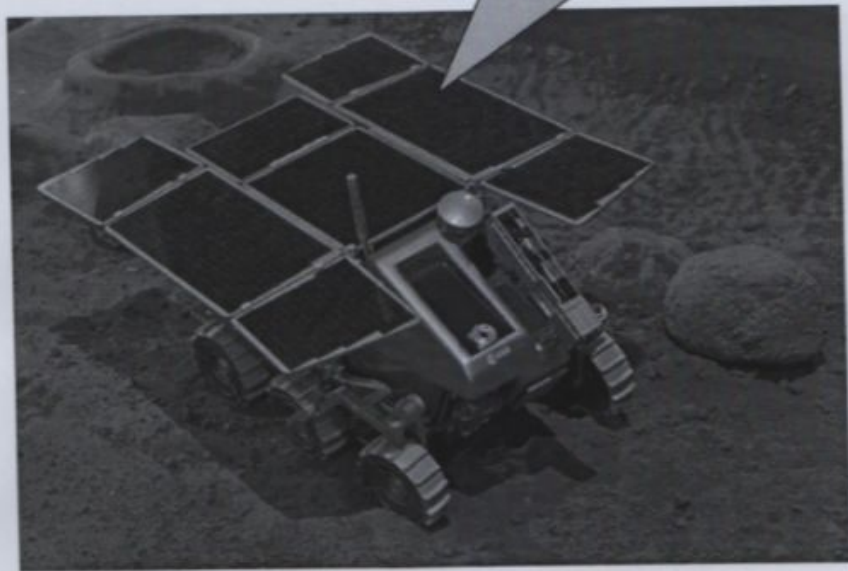


$$\begin{array}{r} 32 \\ \times 12 \\ \hline 64 \\ 320 \\ \hline 384 \end{array}$$

32

Energy per second = 384 J

Area of the panels =  $2.5\text{m}^2$



Each second  $400\text{J}$  of energy from the Sun reaches  $1\text{m}^2$  of the solar panel.

Calculate the total amount of energy reaching the solar panel in 1 second.

[3]

400	400	200
$1\text{m}^2$	$1\text{m}^2$	$0.5\text{m}^2$



The total power input to the leaf blower is 750 W.  
The useful power output of the leaf blower is 360 W.

Calculate the efficiency of the leaf blower.

Use the correct equation from the Physics Equations Sheet.

[2 marks]

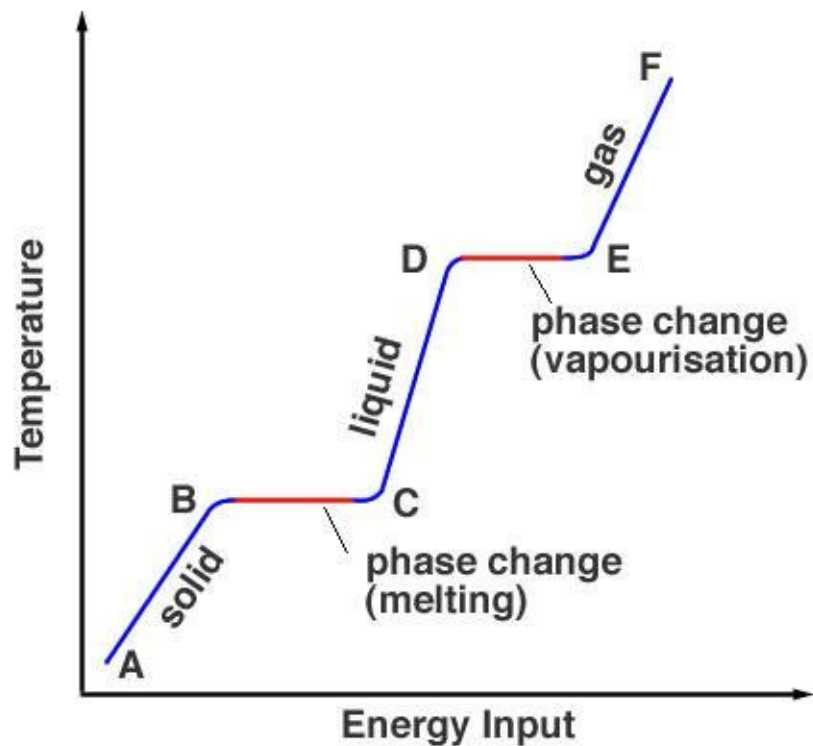
750 W input

360 W - ?

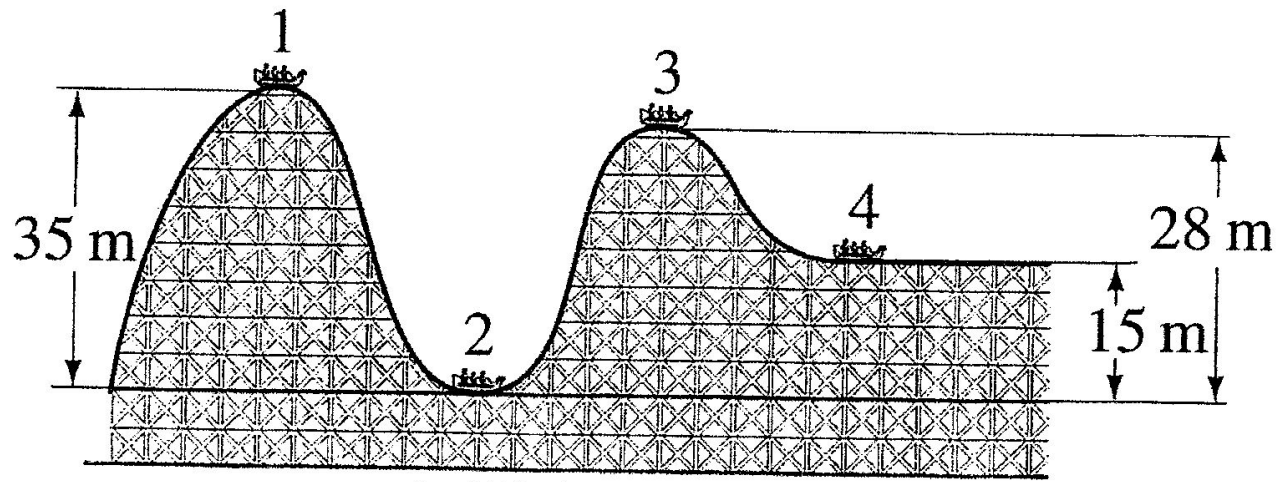
$$\frac{360}{750} \times 100\% = 48\%$$

Efficiency = 48%

# Latent Heat



total energy				
A-B	B-C	C-D	D-E	E-F



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1	GPE			
2	KE			thermal
3	GPE		KE	thermal
4	GPE		KE	thermal

$$M = F \times d$$

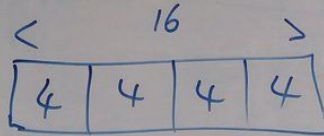
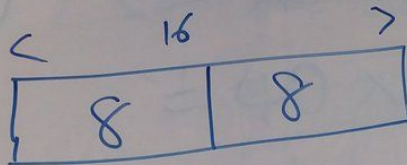
$$\begin{aligned} M &= F \times d \\ &= 4 \times 3 \\ &= 12 \text{ Nm} \end{aligned}$$

$$\begin{aligned} M &= F \times d \\ 12 &= 6 \times ? \\ d &= 2 \text{ m} \end{aligned}$$



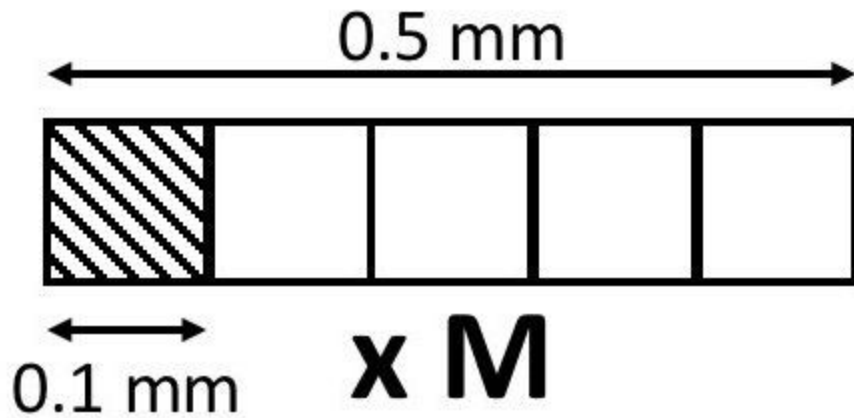
$$\begin{aligned} M &= F \times d \\ &= 8 \times 2 \\ &= 16 \end{aligned}$$

$$\begin{aligned} M &= F \times d \\ 16 &= 4 \times ? \\ d &= 4 \text{ m} \end{aligned}$$



**David Paterson**

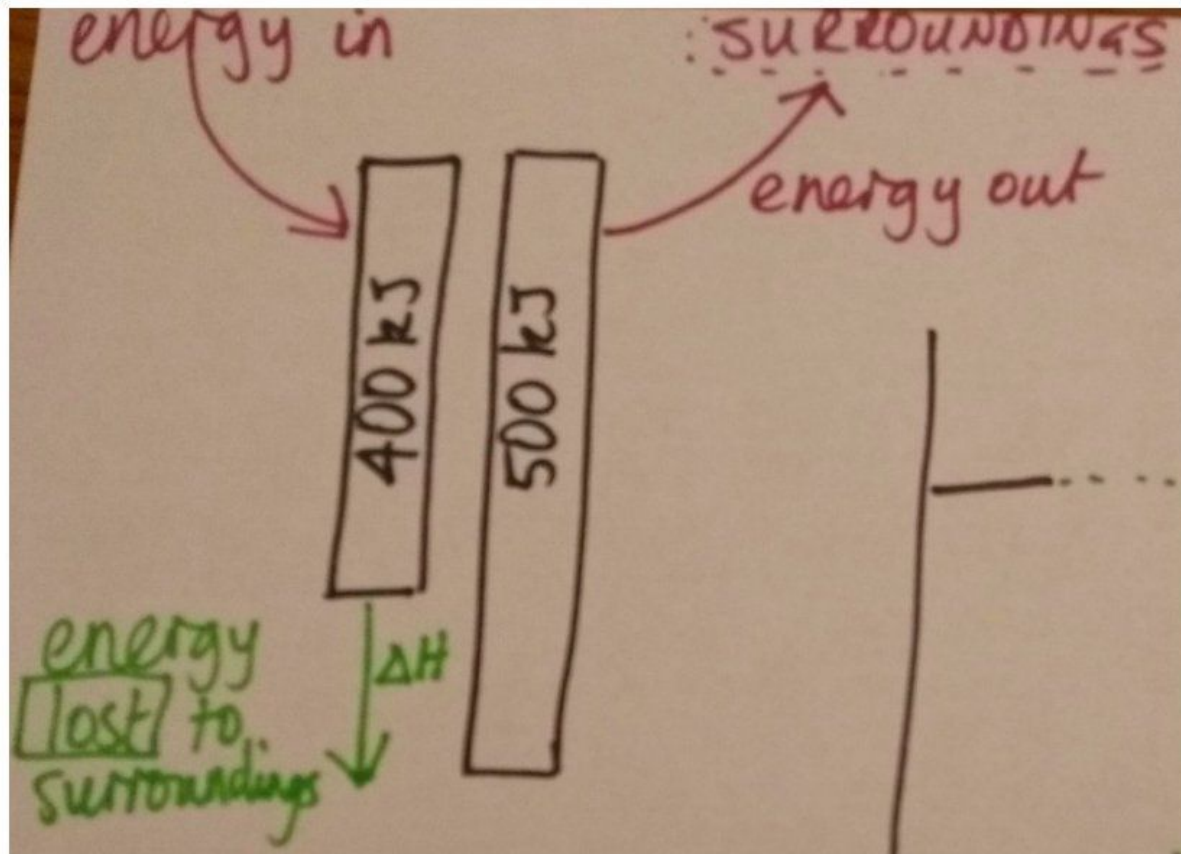
@dave2004b Follows you



Magnification



# Energy, bars and threshold concepts



**Niki Kaiser (K Chem)**

@chemDrK Follows you





**Jenny Koenig** @JennyAKoenig · Mar 15

I've used bar models when using ratio and proportion in chemistry - especially dilutions. I agree completely about transferring maths pedagogy into science!



2



2



4



**Ben Rogers** @BenRogersEdu · Mar 17

Could I take a look at how you did this Jenny? Thanks



1



**Jenny Koenig**

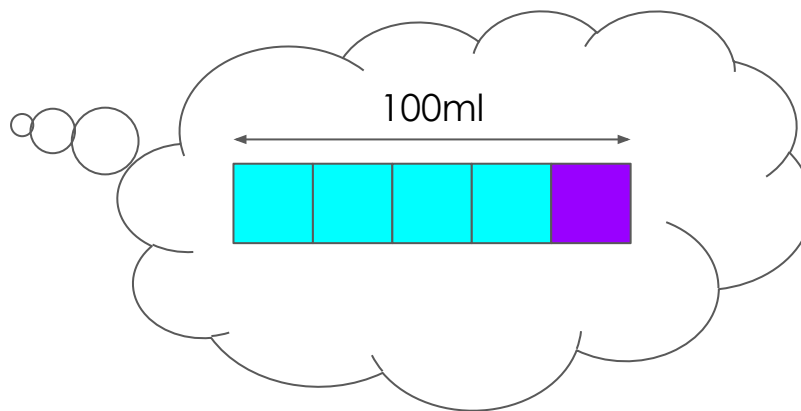
@JennyAKoenig

Following

Replying to @BenRogersEdu @DSGhatura and 2 others

I just sketched diagrams on the board so don't have a handout. In the instructions for diluting ribena it says 1 part to 4 parts water so you can draw that as a bar model then say you want 100ml altogether you can use the bar to work out the volumes.

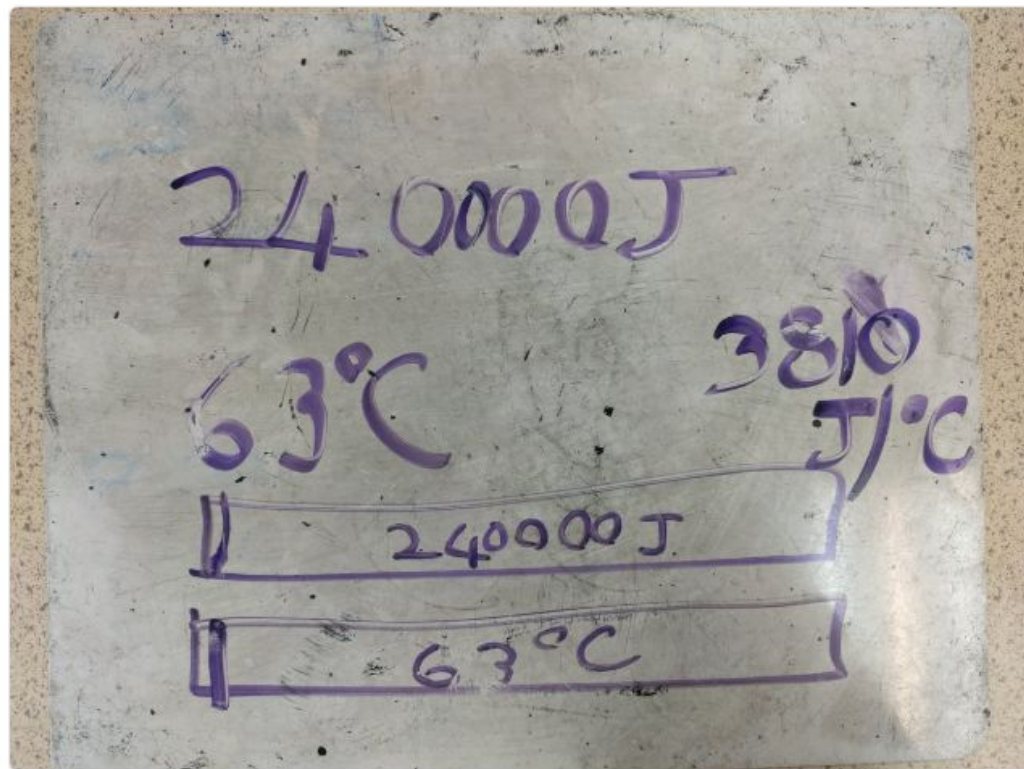
8:17 PM - 18 Mar 2018



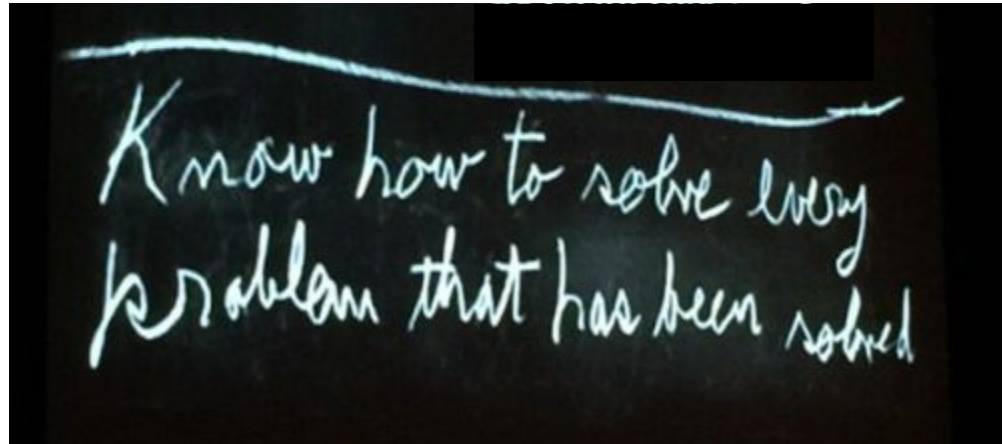


Ruth Smith @ruthyie · Apr 27

@BenRogersEdu Heated bubble bath, wanted to find the energy supplied per degC. (teaching SHC drip by drip to Y9) I tried the bar method you write about to groups and they gasped at the notional 1/63 slice stage and knew what to do with 240 000 J. Seems to help #asechat





A photograph of a chalkboard with a single line of handwritten text in white chalk. The text is written in a cursive, slightly slanted script. Above the text is a long, horizontal, slightly wavy line drawn with the same chalk. The background is a dark, textured surface, typical of a chalkboard.

Know how to solve every  
problem that has been solved

Feynman

A physicist is the sum of  
the problems she can  
solve.

Ben Rogers

